

Regioselective Proton Abstraction and 1,3-Migration of a Phosphorus Group in 1,3-Dienes by Iron Coordination: A New Method for the Synthesis of α -Phosphono- α,β -unsaturated Ketones

Tatsuo Okauchi,* Takao Teshima, Keishi Hayashi, Nobuo Suetsugu, and Toru Minami*

Department of Applied Chemistry
Kyushu Institute of Technology
Sensui-cho, Tobata, Kitakyushu 804-8550, Japan

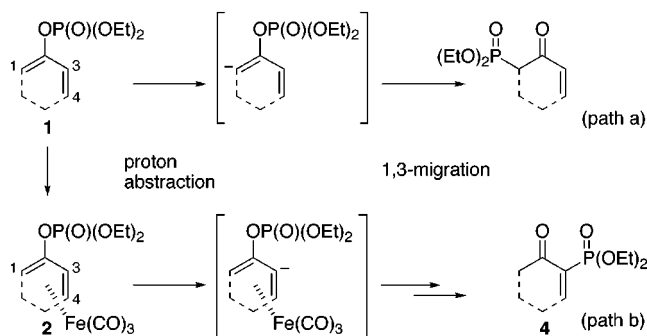
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Vinylphosphonates containing various functional groups have been widely studied due to their synthetic usefulness.¹ We previously reported that vinylphosphonates having an electron-withdrawing group at the α -position underwent Lewis acid-catalyzed cyclizations, for example, intramolecular ene reactions,² [2 + 2] cycloadditions,³ and Nazarov cyclizations.⁴ Our interest in vinylphosphonate chemistry led us to explore the synthesis of cyclic vinylphosphonates bearing an electron-withdrawing group, that is, α -phosphono- α,β -unsaturated cyclic ketones. Although a few reports on the preparation of such kinds of cyclic vinylphosphonates have been found, the preparation procedures are considered to be too lengthy.⁵ Thus, the need for a convenient and general method for the preparation of the cyclic vinylphosphonate is manifest.

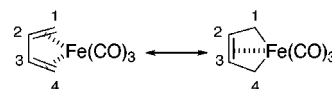
1,3-Migration of phosphorus from oxygen to carbon is convenient for preparing β -keto phosphonates (path **a** in Scheme 1), since dienyl phosphates are easily obtained and rearranged regioselectively to the C-1. This reaction involves regioselective proton abstraction at the C-1 of dienyl phosphate **1** and subsequent migration of phosphorus to give β -keto phosphonate. We envisioned that the migration might be used for the preparation of α -phosphono- α,β -unsaturated ketone if regioselective abstraction of the proton at the C-3 in **1** could be performed.

Iron-1,3-diene complexes have a half-sandwich structure and are described by the following resonance forms (Scheme 2). Considering these forms, the complex would have dialkylmetal characters to some extent, and C-1 and C-4 carbons of the 1,3-diene part would have greater sp^3 character than C-2 and C-3 carbons.⁷ It can, therefore, be presumed that when the complex **2** is treated with a strong base the proton abstraction would occur at C-3 and phosphorus would successively migrate to C-3 to give

Scheme 1

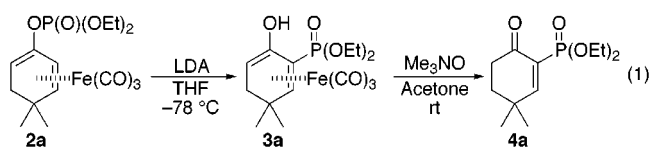


Scheme 2



phosphonoenone **4** after demetalation (path **b** in Scheme 1). We now report the first example of the controlled regioselectivity of the proton abstraction from 1,3-diene by iron coordination⁸ and a convenient method for the synthesis of α -phosphono- α,β -unsaturated cyclic ketones using iron–diene complexes.

Iron complex **2a** was easily prepared from a dienyl phosphate and (η^4 -benzylideneacetone)tricarbonyliron⁹ in 43% yield. Treatment of **2a** with 2.2 equiv of LDA at -78°C selectively furnished 1,3-phosphorus migration product **3a** in 78% yield (eq 1). This



demonstrates that coordination of $\text{Fe}(\text{CO})_3$ to the diene dramatically altered the regioselectivity of the proton abstraction. Removal of the iron moiety from **3a** was accomplished by the oxidation with Me_3NO ,¹⁰ giving the desired α -phosphono- α,β -unsaturated cyclic enone **4a** in 95% yield (eq 1).

Various iron–diene complexes **2b–l** having a phosphate group were prepared from dienyl phosphates **1b–l** and were subjected to treatment with LDA. As summarized in Table 1, the phosphorus migration to C-3 of cyclic and acyclic diene complexes proceeded smoothly to give the corresponding products **3** in good yields (except for **2d**, entry 4). Interestingly, iron–diphenyl dienyl phosphonate complex **2c** was converted to iron-coordinated α -phosphono- α,β -unsaturated ketone **3c** (entry 3). In contrast to this result, Wiemer reported that phosphorus migration to the aromatic ring was observed when diphenyl vinyl phosphate was treated with LDA.¹¹ Therefore, we conclude that acidity of a proton at C-3 of the iron–diene complex is higher than that of an aromatic proton at the ortho position of the oxygen. For acyclic complexes **2i–m**, it is of interest to mention that the elimination was suppressed by iron coordination and that the phosphorus migration to the C-3 occurred exclusively to give iron-coordinated α -phosphono- α,β -unsaturated ketones **3i–m**, since treatment of

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* Corresponding author. E-mail: okauchi@che.kyutech.ac.jp.

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(7) The right form in Scheme 2 is equivalent to the population of back-bonding to the LUMO of butadiene. Back-bonding is not so extensive in metal complexes with carbonyl ligands. However, C-3–C-4 and C-1–C-2 bonds in diene–iron complexes are elongated compared with free butadiene, and pyramidalization angles at C-1 and C-4 are greater than those of C-2 and C-3. These are interpreted that back-bonding in diene–iron complexes is strong enough to increase sp^3 character of C-1 and C-4 higher than that of C-2 and C-3. See, Deeming, A. J. *Mononuclear Iron Compounds with η^2 – η^6 Hydrocarbon Ligands*. In *Comprehensive Organometallic Chemistry*; Wilkinson, G.; Stone, F. G. A., Abel, E. W., Eds.; Pergamon: Oxford, 1982; Vol. 4, p 377.

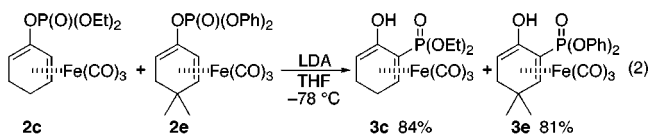
Table 1. 1,3-Phosphorus Migration of Iron Complexes **2**^a

entry	iron complex, 2	time (h)	product, 3	yield (%) ^b
1	2a 	1	3a	quant.
2	2b 	2	3b	64 ^c
3	2c 	1	3c	87
4	2d 	2	—	—
5	2e 	1	3e	quant
6	2f 	0.5	3f	96
7	2g 	1	3g	89
8	2h 	1	3h	quant.
9	2i 	1.5	3i	quant.
10	2j 	1	3j	96
11	2k 	1	3k	92
12	2l 	1	3l	88
13	2m 	1	3m	93

^a Iron complex **2**, LDA (2.2 equiv) were reacted in THF at $-78\text{ }^{\circ}\text{C}$.^b Isolated yield. ^c Reaction run at $-63\text{ }^{\circ}\text{C}$.

iron-free acyclic dienyl phosphates with LDA produces enynes with an elimination of the diethyl phosphoric group.¹²

To determine whether this migration occurred inter- or intramolecularly, a crossover experiment was conducted (eq 2).



Treatment of an equimolar mixture of **2c** and **2e** with LDA only provided products **3c** and **3e**, indicating that the migration is an intramolecular process.

A similar reaction of several iron–diene complexes bearing a sulfonyloxy or benzyloxy group was investigated. The results are

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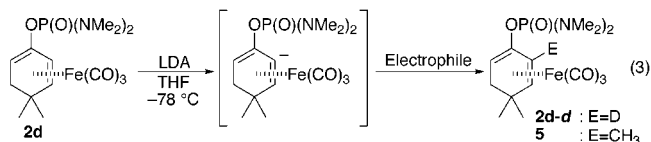
Table 2. 1,3-Migration of a Sulfonyl or Benzyl Group in Iron Complex **5**^a

entry	5 , Y	temp ($^{\circ}\text{C}$)	time	yield of 6 (%) ^b
1	5a , $\text{CF}_3(\text{CF}_2)_3\text{SO}_2$	-78	5 min	80
2	5b , CF_3SO_2	-78	5 min	83
3	5c , PhCO	0	0.5 h	96 ^c

^a Iron complex **5**, LDA (1.1 equiv) were reacted in THF at $-78\text{ }^{\circ}\text{C}$. ^b Isolated yield. ^c **7** was obtained.

shown in Table 2. The migration proceeded quite successfully to give the expected iron complexes **6** or the corresponding decomplexed enones **7**. When the complex with a migrating group having acidic protons, such as a methanesulfonyl or acetyl group, was employed, no migration product was obtained; instead, iron–dienol complex **5** ($\text{Y} = \text{H}$) was formed quite rapidly even at low temperature.

Next, we tried to trap the anion intermediate of the iron–diene complex by an electrophile intermolecularly.¹³ Complex **2d** was selected as the starting substrate because its phosphonic amide group did not migrate (Table 1, entry 4). **2d** was treated with LDA under the same conditions as shown in Table 1, and then D_2O was added into the reaction mixture at $-78\text{ }^{\circ}\text{C}$ (eq 3). ¹H



NMR integration of recovered **2d** (82%) showed 93% deuterium incorporation at C-3. An attempt to trap the generated anion by iodometane afforded the alkylated complex **8** in 76% yield. These are the first examples of generation of the anion from an iron–diene complex and of its reaction with an electrophile.

In conclusion, iron coordination to 1,3-diene with a phosphate group allows proton abstraction at C-3 and subsequent 1,3-migration or alkylation to occur. Removal of the iron moiety from the migration product gives α -phosphono- α,β -unsaturated ketones. This procedure provides a new approach to synthesize cyclic vinylphosphonates bearing an electron-withdrawing group. Furthermore, iron–diene complexes with a sulfonyl or an acyl group are also successfully employed in this reaction. These results indicate that the acidity of protons at C-2 and C-3 of the iron–1,3-diene complex may be higher than that of the C-1 and C-4 protons.

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Supporting Information Available: Detailed experimental procedures including spectroscopic and analytical data (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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